

Appendix B Workshop Abstracts

Table of Contents

Abstracts

The Mercury Marketplace: Sources, Demand, Price and the Impacts of Environmental Regulations	23
Waste Minimization and Elimination	24
EPA's Mercury Action Plan	25
Disposal of Mercury Waste and Stockpiles	26
Mixed Waste Issues	27
International Perspective	28
Mercury Information Management Issues	30
National Implementation of the Universal Waste Rule for Mercury Lamps (Industry Perspective)	31
State Perspective	33
Model State Legislation	34
NGO Perspective	35
Mercury Stock Management	37
Sub-Seabed Emplacement: Long-Term Ultimate Disposition of Mercury Wastes	38
Case Study of a Pilot Scale System for Removal of Organic Mercury from Pharmaceutical Wastewater	39
SAMMS Technology	40
Mercury Collection Programs in Sweden	41
Phase-out of Mercury-Containing Products	43
Demonstration of Mercury Treatment Technologies to Meet DOE Customer Needs	44
Return and Recycling of Used High Intensity Bulbs for Recycling and Closed-loop Mercury Control	46
Mercury Amalgamation Demos with the DOE	47
Deployment of the Sulfur Polymerization and Stabilization Process as Applied to Mercury Contamination in Soils	48
Commercializing a Safer Substitute for Mercury	49
The Business of Mercury Pollution Prevention: Identifying Source Reduction Opportunities and Engineering Trade-Offs	50
A PBT Technology Information Clearinghouse	52

Mercury Stabilization in Chemically Bonded Phosphate Ceramics	55
Characterization and Leachability of Stabilized Mercury-Containing Wastes	57
Treatment of Wastes Contaminated with Mercury	58
Treatment of Mercury-Bearing Wastes with Thermal Desorption Technology	59
Permanent Mercury Disposal in Sweden	60
Sub-Seabed Emplacement: Long-Term Ultimate Disposal of Mercury Wastes in Geologic Formations on Land	62
Mercury-Sniffing Dogs: The Swedish Experience	63
Mercury Source Reduction and Recycling in Electrical Products	64
DSCP Buying Green	66
EPA/AHA Agreement: Reduction of Mercury Wastes from Hospitals/Health Care Facilities	67
Mercury Content of Products Commonly Used by Boston Area Hospitals	68
Eliminating Non-Essential Mercury Uses	69

The Mercury Marketplace: Sources, Demand, Price and the Impacts of Environmental Regulations

Bruce Lawrence
President
Bethlehem Apparatus Company, Inc., 890 Front Street
Hellertown, PA 18055
Phone: (610) 838-7034, Fax: (610) 838-6333
brucelawr@aol.com

Bruce Lawrence

Mr. Lawrence is the President of Bethlehem Apparatus Company, Inc., since 1980, and the principal stock holder since 1992. Bethlehem Apparatus Company is the leading company supplying mercury to the U.S. domestic market, as well as the leading mercury retort recycling operation. Mr. Lawrence has been published in the Engineering and Mining Journal for several years in the annual mineral section on the Mercury Market. He has also presented work to EPA on the retort distillation of mercury, 1992.

The Mercury Marketplace: Sources, Demand, Price and the Impacts of Environmental Regulations

Presentation will provide answers to the following questions. Where does the present market for mercury get its supplies? How does recycling of mercury waste materials effect the market? What is byproduct mercury and how does it interact with the more traditional supplies of mercury? Mercury Mining; Who does it and is it still necessary for the supply to the mercury marketplace? Who still uses mercury in products and services? How is mercury used in consumer products? How is mercury used in non-consumer products? How much mercury is in use today? How much mercury is available for the mercury marketplace? Who owns this mercury? Why are there stockpiles of mercury? What changes have taken place in the past few years since efforts have been made to limit mercury use? How much does mercury cost? Has this changed since the onset of environmental regulation? How does price affect the supply and demand of mercury? Are there other effects of mercury pricing?

Waste Minimization and Elimination

Harold Charles
Waste Minimization and Elimination
U.S. Environmental Protection Agency, Office of Solid Waste
401 M Street, SW
Washington, D.C. 20460
Phone: (703)308-8918, Fax: (703)308-8433
charles.harold@epa.gov

Harold Charles

B.S. in Civil Engineering, University of DC, 1986
M.S. in Waste and Environmental Management, University of MD, 1994
Professional Engineering License with DC and MD, 1997

1987 to 1994, Civil Engineer and Environmental Coordinator, DC Air National Guard at Andrews Air Force Base, 113th Civil Engineering Squadron
1994 to 1995, Environmental Protection Specialist, Headquarters U.S. Army at the Pentagon, Environmental Programs Directorate, Pollution Prevention Division

1995 to 1998, Civil Engineer and Environmental Officer, Headquarters Federal Emergency Management Agency, Response and Recovery Directorate, Infrastructure Division, Engineering Branch

1998 to Present, Environmental Engineer, Headquarters Environmental Protection Agency, Office of Solid Waste, Hazardous Waste Management and Minimization Division, Waste Minimization Branch

Mercury is one of the PBT (persistent, bioaccumulative and toxic) chemicals that EPA has focused on over the years.

An overview will be given of how mercury in products and production process is found in waste streams. Subsequently national data of mercury bearing wastes and how they are managed (i.e. treated, recycled, and disposed of) will be highlighted.

Current EPA initiatives focusing on mercury in wastes will be discussed, including pollution prevention initiatives.

EPA/OSW supports waste minimization to reduce mercury in wastes and when not feasible, effective treatment or more Land Disposal Restriction (LDR).

EPA's Mercury Action Plan

Greg Susanke
U.S. Environmental Protection Agency, Office of Pollution Prevention and Toxics
401 M Street, SW
Washington, DC 20460
Phone: (202)260-3547
susanke.greg@epa.gov

Greg Susanke

Greg Susanke is a biologist, and the manager of EPA's Office of Pollution Prevention and Toxics' Mercury Program. He is currently leading a multi-Office workgroup effort in developing a Mercury Action Plan for the Agency. Greg is also serving as a U.S. representative on the Commission for Environmental Cooperation's Mercury Task Force where he has helped implement Phase I of the North American Regional Action Plan on Mercury, and has assisted in the drafting of its second phase.

EPA's Mercury Action Plan: An Overview

Among the many pollutants that EPA addresses, persistent, bioaccumulative and toxic (PBT) substances are pollutants of primary concern. It has traditionally addressed these and other pollutants among its single-media offices. However, many pollutants, especially PBTs, can not be fully addressed in this manner because of their cross-media nature. Accordingly, the EPA is committing, through the development of a PBT Strategy, to create an enduring cross-office system that will address the cross-media issues associated with priority PBT pollutants. The PBT Strategy, which is currently being drafted, will integrate the work being done across media offices and between national and regional programs more thoroughly. It will align domestic and international activities more effectively, involve stakeholders, and use measurable objectives and assess performance. This strategy is intended to make the whole of the Agency's efforts on PBT pollutants more than a sum of its parts.

A central element to EPA's PBT Strategy is the development and implementation of national action plans for priority PBT pollutants. Mercury has been selected as one of the first PBT substances to be addressed under this strategy. The Agency is currently in the process of drafting a Mercury Action Plan. Although the PBT Strategy will not be discussed in the presentation, an understanding of its principles, as previously mentioned, frame the context of the action plans.

The presentation will briefly discuss use, release, and risk reduction goals for mercury, as well as the tools to be used to measure progress in achieving these goals. A listing of the numerous source categories/sectors to be addressed will be presented, but the focus of the presentation will be on describing the priority areas

of future action. Mention of these priorities at the time of writing this abstract is not possible, as they are currently being developed.

Disposal of Mercury Waste and Stockpiles

Josh Lewis
United States Environmental Protection Agency
Office of Solid Waste (5302 W)
401 M Street, SW
Washington, D.C. 20460
Phone: (703) 308-7877, Fax: (703) 308-8433
lewis.josh@epamail.epa.gov

Josh Lewis

Josh Lewis is an Environmental Engineer in the Waste Treatment Branch of EPA's Office of Solid Waste. He graduated from Cornell University with a B.S. in Environmental Engineering. Josh has worked at EPA for two years, during which time one of his main projects has been the reevaluation of the Land Disposal Restriction treatment standards for mercury-bearing wastes.

Treatment and Disposal of Mercury Hazardous Waste

The original Land Disposal Restriction (LDR) treatment standards for mercury-bearing wastes were promulgated in 1990. These standards, which are still in place, require high mercury subcategory wastes (i.e., wastes that contain greater than or equal to 260 ppm total mercury) to be roasted or retorted to recover the mercury or, if organics are present, the wastes can also be incinerated. Low- mercury subcategory wastes (i.e., wastes that contain less than 260 ppm total mercury) have to meet a numerical treatment standard based on the toxicity characteristic leaching procedure (TCLP). Since 1990, many issues have arisen with the mercury treatment standards, including whether the original premise of incineration as a pretreatment step to mercury recovery is still true; whether there are options for treating high-mercury wastes that are not amenable to retorting; and, since mercury use in industry is on the decline, whether we should still require mercury recovery for high subcategory wastes, or instead allow treaters the option of stabilizing these wastes. Because of these and other issues, EPA has begun a reevaluation of the LDR mercury treatment standards. The first step in this reevaluation was the publication of an Advance Notice of Proposed Rulemaking (ANPRM) on May 28, 1999, which described the issues we have with the current mercury treatment standards and discussed some potential options for amending the standards. We are now evaluating the comments that we received on this ANPRM. In addition, we are involved in two treatability studies that are researching the efficacy of emerging mercury treatment technologies. The end result of our current mercury work will be the publication of a proposed rule on changes to the LDR mercury treatment standards in late 2000.

Mixed Waste Issues

Grace Ordaz
Chemical Engineer
U.S. Environmental Protection Agency
Office of Solid Waste
401 M Street, SW (MC 5304W)
Washington, DC 20460
Phone: (703) 308-1130, Fax: (703) 605-0744
Ordaz.Grace@epa.gov

Greg Hulet
Mixed Waste Focus Area
Bechtel BWXT, LLC
Idaho National Engineering and Environmental
Laboratory
P.O. Box 1625 MS 3875
Idaho Falls, ID 83415-3875
Phone: (208) 526-0283, Fax: (208) 526-1061
Hag@inel.gov

Grace Ordaz

Ms. Ordaz has been working on EPA mixed waste proposal for the past two years. Ms. Ordaz has also worked at US DOE Office of Environmental Management, Office of Research and Development on mixed waste technology development, and at the MD Department of Environment administering the State Biomonitoring Program under CWA's municipal NPDES permit program. Ms. Ordaz also has experience with the AA county pretreatment program under the CWA, and process design of petroleum plants.

Greg Hulet

Mr. Hulet is the work package manager for the Unique Waste Work Package, which includes DOE's mercury mixed wastes. As such, he coordinates research, development, and technology deployment activities to ensure that all the wastes in the Unique category have a path for treatment and disposal. He has a Masters Degree in Chemical Engineering and ten years experience in waste management and pollution prevention. He also has considerable experience with Naval Nuclear Propulsion Plants. He has been a scoutmaster for 15 years, which, after watching scouts cook for that long, has made him an expert in unique hazardous wastes.

EPA Proposed Rule for Storage, Treatment, Transportation, and Disposal of Mixed Waste

Conservation and Recovery Act (RCRA) to provide a conditional exemption from certain requirements for eligible mixed waste. EPA is requesting public comments on this proposed action.

Mixed waste is a radioactive RCRA hazardous waste. It is regulated under two authorities: 1) the Resource Conservation and Recovery Act (RCRA), as implemented by EPA or authorized states for the hazardous waste component; and 2) the Atomic Energy Act of 1954, as amended (AEA), for the radiological component as implemented by either the Department of Energy (DOE), or the Nuclear Regulatory Commission (NRC) or its Agreement States.

The focus of this proposal is to provide flexibility under RCRA Subtitle C to generators of eligible mixed waste. We are proposing a conditional exemption from the definition of hazardous waste applicable to: low-level mixed waste (LLMW) for storage; and LLMW or Naturally Occurring and/or Accelerator-produced Radioactive Material (NARM) for transportation and disposal. The proposal is expected to reduce dual regulation for generators in the management and disposal of their wastes. This flexibility will enable generators of LLMW who are licensed by the Nuclear Regulatory Commission (NRC) to claim an exemption for storing and treating these wastes in tanks or containers (using solidification, neutralization, or other stabilization processes) without a RCRA permit. This proposal will also provide flexibility for the manifesting, transportation and disposal of eligible mixed waste. Waste meeting the proposed conditions will be exempted from certain RCRA Subtitle C hazardous waste requirements and managed as radioactive waste in accordance with NRC regulations.

International Perspective

John Diamante
U.S. EPA, Office of International Activities
401 M Street, SW
Washington, DC 20460
Phone: (202)564-6608, Fax: (315)475-9351
vdemarchi@secor.com

Marilyn E. Engle
U.S. EPA, Office of International Activities
401 M Street, SW
Washington, DC 20460
Phone: (202)564-6472, Fax: (202)565-2409
engle.marilyn@epa.gov

John M. Diamante

John M. Diamante is the Senior Science Advisor for the EPA Office of International Activities, reporting to the Assistant Administrator and Deputy. His responsibilities are to provide advice, review and oversight on technical and scientific matters and related policy issues concerning the programs and activities of the Office. He is actively engaged in interagency and international cooperative projects concerned with radioactive waste management problems in Northwest Russia. He received his doctorate from New York University based on research in planetary atmospheres conducted at the NASA Goddard Institute for Space Studies in New York. Subsequently, he was employed at several aerospace companies, including TRW Systems and EGandG, and then went on to federal employment with the National Oceanic and Atmospheric Administration (NOAA). At NOAA, he served as a scientific and technical advisor in the National Ocean Service, Oceanic and Atmospheric Research Office and Climate Change Program Office.

Marilyn E. Engle

Marilyn E. Engle is an International Affairs Specialist in the EPA Office of International Activities. She presently is the Agency lead on international transboundary transport aspects of mercury, and has served as lead on international marine and coastal issues, where she initiated Agency activities to shape a Land-Based Sources of Pollution (LBS) Protocol for the Wider Caribbean. She received her BA in Zoology and

Anthropology from Duke University and her Master's Degree from George Washington University. Her experiences include being a Senior Research Technician at Duke University Medical Center working conducting research on non-ionizing radiation sublethal effects. After joining EPA, she was an Environmental Scientist for the Office of Radiation Programs and supported the regulatory program on ocean disposal of low-level radioactive waste before taking her present position in the Office of International Activities. She also co-managed the Arctic Nuclear Waste Assessment Program (ANWAP) while on a recent detail from EPA to the Department of Defense Office of Naval Research, where she focused on preparing a human and ecological risk assessment of the potential for transport from Russia to the U.S. State of Alaska of Russian nuclear wastes dumped into or entering the Arctic Ocean.

Long-Range Transboundary Transport of Mercury: International Dimensions of the Mercury Problem and Opportunities for Cooperative Solutions

We are becoming increasingly aware that we must address mercury, a persistent and bioaccumulative toxic, at local, regional and global scales. In addition to the problem of long-range transport from combustion sources of mercury, such as coal burning, the EPA Office of International Activities (OIA) also sees a need to focus on the long-range transport from non-combustion sources, such as the chlor-alkali industry and mercury in waste streams.

There is growing evidence that the U.S. is being impacted by many atmospherically borne, globally circulating persistent toxics, such as persistent organic pollutants (POPs), and other atmospheric contaminants, including ozone and particulates. There is reason to believe that mercury is similarly being transported to the U.S. from abroad, and that U.S. sources are contributing to the global pool of mercury that is being circulated worldwide. EPA estimates that about one-third of U.S. anthropogenic mercury emissions are deposited in the contiguous U.S., while the remaining two-thirds is transported outside the U.S. and enters the global pool. Correspondingly, estimates suggest that about 35 tons, or 40% of the mercury that is deposited in the U.S. per year, may originate from sources external to the U.S. With the rapid industrialization and increasing use of coal in Asia, and re-industrialization in Russia, this trend is expected to increase. Rapid industrialization will also increase the burden arising from the non-combustion sources.

The mission of OIA regarding mercury is multifold: 1) to improve understanding of international sources of mercury, and the regional and global-scale transport processes; 2) to influence international awareness and actions through international fora; 3) to provide international training and technology transfer in selected countries to bring about reductions in mercury use and emissions; and 4) to facilitate data and information management. Our emphasis to date has been on improving scientific understanding of long-range transport, and on partnering with other countries in cooperative solutions, and through regional fora to collectively influence actions in other countries. Currently, OIA, in cooperation with other EPA Offices, other federal agencies and other governments, is supporting activities such as speciated mercury monitoring and modeling efforts in Barrow, Alaska and in the Ohio River Valley and the Florida Everglades to evaluate international contributions of mercury to U.S. deposition. EPA is also actively engaged in mercury issues

and regional action plans under numerous regional agreements, including the U.S.-Canada Binational Toxics Strategy, the North American Commission on Environmental Cooperation (CEC) involving the U.S., Canada and Mexico, the UNECE Convention on Long-Range Transboundary Air Pollution (LRTAP) Heavy Metals Protocol, and the activities of the Arctic Council, which includes the Arctic Monitoring and Assessment Program (AMAP).

In addition to improving scientific knowledge of transport and fate of mercury sources, we are working through international fora to find opportunities for international cooperative approaches to further: 1) source identification and characterization, particularly with the chlor-alkali sector; 2) pollution prevention, such as taking mercury out of products; 3) environmental capacity building; 4) environmentally sound trade and free market decisions regarding mercury, and 5) informed international policy making concerning mercury.

Mercury Information Management Issues

James Ekmann
Office of Systems and Environmental Analysis,
National Energy Technology Laboratory,
U.S. Department of Energy
Phone: (412)386-5716
ekmann@netl.doe.gov

James Ekmann

Mr. Ekmann serves as the Deputy Associate Director in the Office of Systems and Environmental Analysis. This office is part of the National Energy Technology Laboratory of the U.S. Department of Energy. OSEA assesses the technical, environmental, and cost performance of technologies developed at or under funding from NETL. Staff in the office conduct environmental assessments, detailed engineering reviews in support of RDandD projects. The office also provides a focal point for the laboratory's external communication including technology transfer, and preparation of materials summarizing technical successes.

Information Tools for Mitigation Strategy Development

The need to link technology costs and a comprehensive risk assessment methodology in the context of addressing major environmental contaminants, e.g., mercury and other persistent bioaccumulative toxics (PBTs) has been discussed by a number of authors. Assessments of policy options rely increasingly on multiple tiers of modeling studies informed by large volumes of data. This tendency raises the need to manage

the use of models and the data needed to ensure analytical results that are consistent and of sufficient quality. The NETL has been examining the connection between data, and concepts such as information, knowledge, and wisdom as these relate to the role of advanced fossil fuel technologies in a carbon managed future. We plan to develop a decision support model that would be an information portal to both process-level data and information and to system-level analyses. We believe that this linkage will lead to knowledgeable choices about mitigation technologies and has the potential to clearly communicate results facilitating formulation of wise policy options. We believe that this endeavor offers useful insights to similar information needs and structures for other issues such as mercury /PBTs. This paper discusses both the approach being used to design the decision support system and the linkages between scientific and technical data and information on societal values that are essential to making such a concept useful.

National Implementation of the Universal Waste Rule for Mercury Lamps (Industry Perspective)

Paul W. Abernathy
Executive Director
Association of Lighting and Mercury Recyclers
2436 Foothill Blvd. Suite K
Calistoga, CA 94515
Phone: (707) 942-2197, Fax: (707) 942-2198
abernath@napanet.net

Paul W. Abernathy

Paul W. Abernathy is the Executive Director of the Association of Lighting and Mercury Recyclers, a national non-profit organization representing members of the mercury recycling industry. Mr. Abernathy has worked for over 25 years in the environmental services industry throughout North America, representing public and private companies and clients. His background includes extensive participation in public policy development and implementation for water quality, air and hazardous substances management. Mr. Abernathy has had experience working with NATO on international exchange of environmental management programs and technologies; was appointed by a California governor to serve on the multi-disciplined State Hazardous Waste Facility “Siting” committee; and presently serves as technical advisor to regional governments in Northern California on hazardous waste management planning, siting and development issues, water and energy conservation, regulatory and environmental compliance, pollution prevention and resource recovery. Paul serves as member of Northern California Green Business Advisory Board.

Mr. Abernathy earned a M.B.A. from Pepperdine University and a B.S. in Biology/Chemistry from the University of Arizona.

National Implementation of the Universal Waste Rule for Mercury Lamps

This presentation includes a brief history of mercury lamp recycling and disposal in the U.S., the public policies that have influenced lamp disposal, highlights of states' programs regulating lamps, and it discusses the latest changes to the Universal Waste Rule effective 1/6/00. EPA's goal is to divert mercury lamps from municipal wastes, and the Association of Lighting and Mercury Recyclers is part of a public-private partnership that is forming to work with business and all state and local governments in the U.S. for implementation of the new rule. This presentation discusses local government roles and options, business and generator options and the educational and resource information being developed.

Spent mercury lamps are considered hazardous waste, but for the most part they have not been managed this way. EPA believes the major reason for the wholesale non-compliance is the lack of awareness and access to information on the part of lamp owners and local governments, which includes nearly everyone. The national recycling rate has been about 12%, which means there are still 500,000,000 mercury lamps disposed in the garbage, potentially exposing people and the environment to mercury. RCRA has always required the proper management of mercury lamps as hazardous wastes, but with few exceptions (MN, FL) there has been little or no enforcement by regulatory agencies. EPA adopted the UWR to include lamps on 7/6/99. (*FR July 6, 1999, Volume 64 Number 128, pp. 36465-36490, and 40 CFR 273*), effective 1/6/00. The goal of the rule is to increase the recycling rate to 80% and remove regulatory and cost burdens for those who recycle. States may take several possible actions to achieve consistency with RCRA. States may have more stringent policies, but the minimum regulatory criteria must not allow the land disposal of mercury-lamps. Local governments have a responsibility too, through their franchises for solid wastes, HHW programs, SQG programs, pollution prevention programs, landfill operations.

Our recycling association has formed a partnership with Earth's 911, and along with EPA and corporate partners is helping provide information and resources to the states, and working with local governments to adopt and implement programs that encourage recycling and set up a sufficient infrastructure to divert mercury lamps from municipal wastes altogether by making recycling easy, inexpensive and available to business and the public.

The new UWR makes it easier than before and less costly to manage lamps properly. Where RCRA has not been enforced and the compliance rates are low, non-compliant disposal has cost little to generators. Proper lamp management under the UWR represents a small percentage of total lighting costs, and it keeps mercury from being released into the environment. To achieve compliance it is incumbent on states and local government agencies, working with both public and private entities, to ease the burden on generators by making collection and recycling programs for mercury lamps readily available. By sharing information, conducting public-private seminars and workshops throughout the country, offering Earth's 911 resource guide, website and toll-free number we are helping educate people about their responsibility. The national goal is to recycle as many mercury lamps as possible.

State Perspective

John Gilkeson
Principal Planner
Minnesota Office of Environmental Assistance
520 Lafayette Rd. N.
St. Paul, MN 55155-4100
Phone: (651) 215-0199, Fax: (651) 215-0246
john.gilkeson@moea.state.mn.us

John Gilkeson

John Gilkeson has worked for the state of Minnesota for 10 years and is currently a principal planner with the Office of Environmental Assistance. During that time John has worked on "problem and special wastes," including medical and infectious waste, household hazardous waste, batteries, lead, electronics, and mercury wastes. John's focus for the past four years has been on the use and management of mercury in products. John has worked on the Minnesota universal waste rule, the federal mercury lamp rule, the federal mercury stockpile issue, and represents Minnesota on the Binational Toxics Reduction Strategy Mercury Work Group. John has also worked with several industries and sectors that use, manage, or release mercury, including oil refineries, thermostat manufacturers, relay manufacturers, automobile manufacturers, the state dental association, demolition contractors, and several mercury recyclers.

Minnesota State Perspective

Minnesota and other states are taking a variety of approaches to understanding, reducing, and managing mercury that is released from a variety of human activities. Though states have differing needs and resources, and must take different approaches, they also have much in common and would benefit from more coordination in laws, rules, programs, and research. Similarly, on a national and international basis, our common interests would benefit from a more coordinated approach to research, programs, and policy. Other public and private sector interests are key players in these processes and have a strong interest in consistent and equitable measures to address mercury nationally and internationally.

In this presentation, Minnesota state agency staff will present their perspective on impediments to and opportunities for advancing local to international mercury reduction efforts in the areas of:

- Environmental research and monitoring;
- Laws and regulations;
- Policies and programs;
- Education;
- Incentives and other measures for voluntary action, including national early reduction credit;
- Coordination among governments, businesses, and NGOs;
- Research and measures for reducing and managing mercury in purposeful use (and waste management);

- Research and measures for reducing emissions from energy and resource sectors;
- Research and measures for reducing emissions from other unintentional use or material reuse;
- Management and disposition of stockpiles and reserves;
- Retirement of mercury removed from commerce; and
- Developing and promoting non-mercury products and processes.

The presentation will include an overview of recommendations from the Minnesota Comprehensive Mercury Reduction Initiative (March 1999) and International Policy Recommendations developed by attendees of the 5th International Mercury Conference in Rio de Janeiro (May 1999).

Model State Legislation

Richard Phillips
Virginia Department of Environmental Conservation
103 South Main Street
Waterbury, VT 05671
Phone: (802)241-3470, Fax: (802)241-3273
rich@dec.anr.state.vt.us

Richard Phillips

Richard Phillips spent two years designing and overseeing construction of water systems on the Navajo reservation. For the last 30 years he has supervised and managed programs for the Vermont Department of Environmental Conservation. Mr. Phillips managed the construction grant program, the wastewater operation oversight program, the enforcement program and the P2/Assistance programs. He has been responsible for the implementation of Vermont's mercury products labeling and disposal ban law passed in 1998. He has been involved with the development of the regional model mercury products legislation.

Mr. Phillips has a Bachelor and Master's degree from Northeastern University.

Model State Legislation (Northeast States Model)

This presentation is based on efforts of the Northeast States to develop model state legislation. This presentation will describe:

1. The basis for creating model mercury product legislation as recommended in the Regional Mercury Action Plan.
2. The process used to develop the model legislation.
3. The high points of the current draft model legislation which includes the following sections:

Legislative Findings	Public Outreach and Education
Definitions	Universal Waste
Interstate Cleaning house	State Procurement
Notifications	Enforcement
Phase-out and Exceptions	State Review
Labeling	Severability Clause
Disposal Ban and Scrap Facilities	Effective Date
Collection	Administrative Fees
Sales Restrictions	Appropriation
Disclosure	Public Notification and Review
Limitations on Use	Prohibition

4. The remaining steps to adoption as a regional model.
5. The status of state-by-state legislative initiatives.

NGO Perspective

Jane Williams
Executive Director
California Communities Against Toxics
P.O. Box 845
Rosamond, CA 93560
Phone: (661) 273-3098, FAX: (661) 947-9793
Danloan@aol.com

Jane Williams

Ms. Jane Williams serves as the executive director of California Communities Against Toxics, a coalition of 80 community based environmental groups in California. She has a degree in economics from the University of California, Los Angeles and has eight years experience working on environmental issues with a focus on persistent, bioaccumulative toxins, Superfund sites, incineration, and nuclear issues.

She has worked extensively with community-based environmental/public health advocacy groups and Native American tribes on numerous pollution-related issues. Ms. Williams has also worked in Mexico on environmental issues with the Secretaria de Relaciones Exterior, the Instituto Nacional de Ecologia, Commission Nacional del Agua, and with non-governmental organizations in Mexico. She has presented papers at three different conferences in Mexico dealing with pollution and water policy issues.

She is also the Chair of the Waste Committee for the National Sierra Club. This committee has responsibility over many of the Club's pollution related issues including Toxic and Nuclear Waste, Superfund, Brownfields, Nuclear and Chemical Weapons, Solid and Medical Waste, Federal Facilities, and Environmental Justice issues related to waste.

Ms. Williams serves on the board of the California Environmental Research Group, the Clean Air Network, Greenaction, the California Stop Dioxin Exposure Campaign, the Del Amo Action Committee, and the Nonstockpile Chemical Weapons Forum. She is a past member of the Federal Advisory Committee on the Industrial Combustion Coordinated Rulemaking and a former member of the Regulatory Structure Update Technical Advisory Committee on Superfund for the State of California Department of Toxic Substance Control.

NGO Perspective

The United States and Canada agreed to the virtual elimination of persistent toxic substances into the Great Lakes under Article II of the Great Lakes Water Quality Agreement signed November 18, 1987. The current Mercury Action Plan does not serve as an integrated blueprint for actions that will achieve the elimination of mercury emissions into the environment. Forty states now have fish consumption advisories for mercury in fresh water fish due to the continued release of mercury into the air and water. Non-governmental organizations have become concerned about the lack of "linkage" between current EPA policy on mercury and the virtual elimination goal. They have set forth a series of recommendations which they believe would lead to the attainment of this goal, including steps that the EPA should take both in the short term and the long term. This paper will present these recommendations along with the rationale for their adoption.

Mercury Stock Management

Folke Dorgelo
Internal postal code 655
Directorate-General for Environmental Protection
Ministry of Housing, Spatial Planning and the Environment
P.O. Box 30 945
2500 GX THE HAGUE
The Netherlands
Phone: + 31 70 339 4908, Fax: + 31 70 339 1297
Folke.Dorgelo@DSVS.DGM.minvrom.nl

Folke Dorgelo

Mr. Dorgelo's role at the Ministry of Housing, Spatial Planning, and the Environment encompasses heavy metals policy, negotiations with the metal industry in The Netherlands – especially concerning the reduction of corrosion and run-off of copper, zinc and lead used for construction and building; recycling of plastics and packaging (waste) containing heavy metals; risk evaluation and risk management of metals (lead, mercury, cadmium, copper, zinc, chromium, nickel, bismuth, tin) and PNAs; and chemicals risk reduction programme of the OECD, Environmental Health and Safety Division (lead, mercury and cadmium). Mr. Dorgelo also participates in the European Commission DG Enterprise working group (chemicals, plastics and rubber) on the 'limitations on marketing and use of dangerous substances and preparations' (Directive 76/769/EEC).

Mr. Dorgelo earned his M.S. in Biochemistry (1974) from the State University of Leiden, has a Teaching Degree in Chemistry and is a registered toxicologist (Dutch Society for Toxicology).

PARCOM Decision 90/3 (1990) aims at the phase-out of the mercury cell process in the chlor-alkali industry in Europe by 2010. About 12,000 tons of mercury in Europe are now in use in this process. It is expected that these mercury stocks from the chlor-alkali industry, when becoming available due to phase-out of the mercury cell process, will end up in worldwide uncontrollable applications with diffuse emissions to air, water and soil. This concern for global transportation, application and emission of mercury is the main reason for the Netherlands to start a project to achieve commitments with industry for an environmentally proper and sustainable handling, transportation and disposal of the mercury stocks.

Mercury Stock Management

Mercury mining in Spain produces about 1,000 tons of mercury per year, mainly for export. No European policy dealing with the primary and secondary flows of mercury exists up to now.

The presentation will focus on the flow of mercury in the Netherlands, including recycling of mercury-containing waste to technical grade mercury. Experiences with two chlor-alkali production plants in the Netherlands phasing out their mercury cell process will be presented.

The actual situation of the mercury stocks in Europe will be presented with some preliminary policy options.

Sub-Seabed Emplacement: Long-Term Ultimate Disposition of Mercury Wastes

Leo S. Gomez, Ph.D.
Sandia National Laboratories
P.O. Box 5800, MS-0779
Albuquerque, NM 87185
Phone: (505)284-3959, Fax: (505)844-2348
lsgomez@sandia.gov

Leo S. Gomez, Ph.D.

Dr. Gomez has worked in nuclear waste management at Sandia National Laboratories in Albuquerque, New Mexico since 1977. He has been the biological research project manager for four ocean disposal projects and works in the Performance Assessment Department for the Waste Isolation Pilot Plant, a transuranic waste repository in southeastern New Mexico. Before going to Sandia, Dr. Gomez worked on a cancer therapy project at Los Alamos National Laboratory in New Mexico, and worked on a project to detect low levels of transuranic elements in workers at Oak Ridge National Laboratory in Tennessee.

Dr. Gomez has served as a U.S. representative on three international ocean pollution commissions. He is also an editor of the multinational journal, *Radioactive Waste Management and Environmental Restoration*. In addition to his work in nuclear waste management, Dr. Gomez has worked with the Institute of Public Policy at the University of New Mexico investigating the public's perceptions of risk of nuclear technologies. He has also been involved with Sandia's educational outreach activities from kindergarten through the college level.

Leo Gomez earned a Ph.D. in Radiation Biology at Colorado State University in 1973.

Emplacement of Mercury Wastes in the Sediments of the Deep-Ocean?

The primary goal of the U.S. Subseabed Disposal Project (SDP) was to assess the technical and environmental feasibility of disposing of high-level nuclear wastes in deep-sea sediments. Subseabed

disposal, like other geological disposal options, was a multibarrier concept that studied the feasibility of burial of solidified and packaged high-level nuclear waste or spent nuclear fuel in high-integrity canisters, tens of meters within the stable geologic formations of the deep-ocean floor. These deep-ocean floor geologic formations are some of the most stable and predictable on earth. In the subseabed concept the multiple barriers of the waste form, the canister, the clay sediments, and the ocean waters were predicted to delay migration of radionuclides until they decayed to innocuous levels.

The SDP was comprised of the following task groups: Site Assessment, Engineering Studies, Near Field, Sediment Barrier, Physical Oceanography, Biological Oceanography, Radiological Assessment, and Legal and Institutional. The SDP research team developed biosphere transport models to predict the oceanic transport of radionuclides. Researchers also developed the capability to determine and evaluate the risks associated both with normal disposal operations and with potential accidents. Safety assessments contributed to evaluation of the feasibility of the subseabed concept and helped focus required work to answer the feasibility questions. Even though the SDP models were developed to predict the transport of radionuclides, they can be used to predict the biosphere transport of non-radioactive environmental pollutants, such as mercury products and other toxic metals. Many of these pollutants cannot be destroyed or broken down through treatment or environmental degradation, and through physical, chemical, or biological processes will ultimately be deposited in the oceans.

Case Study of a Pilot Scale System for Removal of Organic Mercury from Pharmaceutical Wastewater

Patrick J. Cyr
Advanced Geoservices Corp.
Chadds Ford Business Campus
Routes 202 and 1, Bradywine One, Suite 202
Chadds Ford, PA 19317
Phone: (888)824-3992, Fax: (610)558-2620

Funded by:
Wyeth Ayerst Pharmaceutical Company and the
Institute for Environmental Engineering Research, Villanova University

Patrick J. Cyr

Mr. Cyr has worked in the environmental industry since 1995, practicing environmental, civil, and geotechnical engineering. He has served as project/resident engineer for landfill construction projects and remediation of wetlands. His experience in the environmental field includes removal of contaminants from wastewater, compiling and evaluating data from contaminated sites, civil design, and management of a water

quality database. He has conducted lab testing of samples in an environmental and geotechnical laboratory. He also has experience in design, construction, and testing of pilot plants.

Mr. Cyr earned his Masters degree in Civil/Environmental Engineering (1999) from Villanova University, and his Bachelor of Science degree in Civil Engineering (1996) from Worcester Polytechnic Institute.

Case Study of a Pilot Scale System for Removal of Organic Mercury from Pharmaceutical Wastewater

Mercury discharged to the environment puts the public health and the environment at risk for toxic effects. Organic mercury as thimerosal (a benzene mercury sodium salt: $C_9H_9HgO_2SNa$) is used as an antiseptic and preservative in topical medicines, cosmetics, and vaccines. Hospitals use thimerosal for standard lab tests, such as albumin, herpes, hepatitis, and HIV, etc. Thimerosal and trace amounts of Hg^{2+} are present in wastewater from the manufacture of certain pharmaceutical drugs and quality analysis/control procedures. The scope of this study was to examine the technical feasibility of using adsorption technology for removing thimerosal and inorganic mercury from a pharmaceutical wastewater. Several adsorbents were selected based on their physical and chemical properties and their adsorption affinity for mercury. Batch isotherm and column studies were conducted to determine the most suitable adsorbent for removal of mercury. Results showed that F-400 GAC provided the best results for the removal of thimerosal and Hg^{2+} . A pilot plant was designed, constructed, and tested successfully for treatment of wastewater from a pharmaceutical manufacturing facility.

SAMMS Technology

Nick Lombardo
Pacific NW National Laboratory
PO Box 999
Richland, WA 99352
Phone: (509)375-3644
nj.lombardo@pnl.gov

Shas V. Mattigod
Pacific Northwest National Laboratory
PO Box 999
Richland, WA 99352
Phone: (509)376-4311, Fax: (509)376-5368
shas.mattigold@pnl.gov

Self-Assembled Monolayers on Mesoporous Materials (SAMMS): A Novel Adsorbent for Mercury Removal from Waste Streams

A new class of hybrid mesoporous materials has been developed at the Pacific Northwest National Laboratory for removing toxic heavy metals such as mercury from aqueous and nonaqueous waste streams.

The basis of these novel adsorbent materials are organized monolayers of functional molecules covalently bound to a siliceous mesoporous support. The mesoporous supports are synthesized using surfactant liquid crystalline templates. The resulting mesoporous materials have high surface areas and ordered porosity in the nanometer size range. Functional molecules capable of selectively binding of mercury (thiol groups) are covalently attached to the mesoporous substrates as densely populated monolayers. Mercury adsorption data obtained over an eight order range equilibrium concentrations indicated that thiol-SAMMS can achieve Hg loading as high as ~635 mg/g. The high affinity for Hg adsorption by this material was reflected by K_d values as high as 1×10^8 ml/g. The data also showed that mercury adsorption by thiol-SAMMS was not affected by the initial form of Hg (nitrate, chloride, and methylated) in solution. A study of mercury adsorption kinetics indicated that thiol-SAMMS bound Hg rapidly (about 99.9% adsorption occurring within first five minutes). Tests showed that neither the pH (2 to 10) or the ionic strength (0.01 to 4M) of simulated waste solutions did not significantly affect the mercury adsorption capacity of thiol-SAMMS. Waste streams containing Hg also typically contain many other cations (Ca, Cd, Cu(II), Fe(II), Ni, Pb, and Zn) and complexing anions (Cl, CN, CO_3 , SO_4 , and PO_4). Tests were conducted to examine the competitive adsorption effects of these cations, and the complexation effects of anions on Hg adsorption. The results indicated that the mercury adsorption capacity of thiol-SAMMS was not impaired by the presence of these cations and anions that would be present in different types of waste solutions. The reason for this noncompetitiveness of other cations appears to be due to preferential binding of a softer cation (Hg) by thiol functional groups. These adsorption characteristics show that thiol-SAMMS is a versatile and cost-effective material for removing, recovering, and recycling Hg from various types of waste streams.

Mercury Collection Programs in Sweden

Kristina von Rein
Principal Technical Officer
Section for Chemicals Control
Swedish Environmental Protection Agency
S-106 48 Stockholm, Sweden
Tel: +46(0)8-6981127, Fax: +46(0)8-6981222
Kristina.von-rein@environ.se

Kristina von Rein

Ms. von Rein has been with the Swedish Environmental Protection Agency since 1990, and is the project leader for the Governmental Assignment that includes both an Action Programme for more efficient collection of used goods and products containing mercury and preparation of a proposal for final disposal in Sweden of mercury-containing waste.

Ms. von Rein has a M.S. degree in chemical engineering.

Mercury Collection Programs in Sweden

Phase-out of mercury

Several years ago Sweden decided that the use of mercury should eventually cease altogether, the target year being 2000. A mercury phase-out means that it is firstly the input of new mercury to society that is reduced. Still, large quantities of mercury are present in goods and products still in use. It has been estimated that in Sweden alone (8 million people) there are hundreds of tonnes of mercury in circulation in products.

Action Programme for the collection of mercury

The Swedish EPA was engaged in an Action Programme (1994 - 1999) as instructed by the government in order to improve the efficiency of mercury collection. The Swedish state had allocated about 20 million SEK for this purpose. The SEPA has given aid to 49 projects as well as carried out several projects of their own.

The SEPA programme has focussed on increasing the collection of hidden mercury in the form of:

- clinical thermometers containing mercury,
- mercury in technical goods and products,
- metallic mercury on shelves and in storage rooms, and
- “historic” mercury (in sinks, floor drains, tubes, etc.).

Many efforts undertaken in the action programme have been aimed at mercury inventory, on one hand identification and labelling of mercury in use and on the other hand collection of worn out mercury and discarded goods and products containing mercury.

A total of 10 – 11 tonnes of mercury has been identified, 6 – 7 of which have been collected and 3,5 – 4 tonnes have been labelled.

New ways of finding and collecting mercury

In different regions in Sweden, specially trained electricians, so-called mercury detectives, were visiting companies, local businesses, municipal sewerages to identify and collect or label mercury- containing products. Also, some projects involved tracing mercury with the world’s first mercury dogs, Froy and Ville. The dogs have been searching for mercury in schools and at universities, finding mercury while saving both time and money. Several tonnes of mercury have been found this way. Swedish municipalities and county administrative boards have participated in all projects.

Phase-Out of Mercury-Containing Products

Folke Dorgelo
Internal postal code 655
Directorate-General for Environmental Protection
Ministry of Housing, Spatial Planning and the Environment
P.O. Box 30 945
2500 GX THE HAGUE
The Netherlands
Phone: + 31 70 339 4908, Fax: + 31 70 339 1297
Folke.Dorgelo@DSVS.DGM.minvrom.nl

Folke Dorgelo

Mr. Dorgelo's role at the Ministry of Housing, Spatial Planning, and the Environment encompasses heavy metals policy, negotiations with the metal industry in The Netherlands – especially concerning the reduction of corrosion and run-off of copper, zinc and lead used for construction and building; recycling of plastics and packaging (waste) containing heavy metals; risk evaluation and risk management of metals (lead, mercury, cadmium, copper, zinc, chromium, nickel, bismuth, tin) and PNAs; and chemicals risk reduction programme of the OECD, Environmental Health and Safety Division (lead, mercury and cadmium). Mr. Dorgelo also participates in the European Commission DG Enterprise working group (chemicals, plastics and rubber) on the 'limitations on marketing and use of dangerous substances and preparations' (Directive 76/769/EEC).

Mr. Dorgelo earned his M.S. in Biochemistry (1974) from the State University of Leiden, has a Teaching Degree in Chemistry and is a registered toxicologist (Dutch Society for Toxicology).

Phase Out of Mercury-Containing Products in the Netherlands

The pollution by mercury in the Netherlands is largely caused by mercury-containing products. Closer examination of the Dutch flow of mercury into soil shows that in 1990 over 40% of the flow originated from mercury-containing products. For surface water, a similar percentage comes from mercury-containing products and for sewage sludge over 80% originated from mercury-containing products. This shows that taking product-oriented measures makes a relevant contribution to the reduction in mercury emissions and in addition to the quality of sewage sludge. To determine which products contain mercury and which alternatives are available, an inventory research was carried out.

The data from this research partly forms the basis for the Dutch 'Decree on products containing mercury 1998'. The use of mercury in the Netherlands was estimated at 12.5 tonnes in 1994. Approximately 45% of this can be accounted for by the use of amalgam in dental surgeries. Since 1991, emissions into the environment have been greatly reduced by the use of special amalgam separators. Approximately 40% is

used in various measuring instruments, electro-technical products and in lighting. The remaining 15% is used in batteries, chemicals, pharmaceutical preparations and in the chlor-alkali industry (mercury cell process).

The Decree is intended to achieve the mercury emission reduction objective. Through a ban on the manufacture of and trade in products containing mercury where alternatives are available, the supply of mercury within the economic circuit will be reduced by approximately 35%, or 4.3 tonnes per year. This relates to products such as measuring instruments and electro-technical products. As a result of the Decree, mercury emissions will gradually decrease, because it will take a few years before all the products containing mercury which are in use are replaced by mercury-free alternatives. Starting from the Dutch emission levels in 1990, the emission into sewage sludge, soil and water will decrease by 30%, 20% and 15% respectively. Existing facilities are used for the safe disposal of mercury-containing products.

Demonstration of Mercury Treatment Technologies to Meet DOE Customer Needs

Greg Hulet
Mixed Waste Focus Area
Bechtel BWXT, LLC
Idaho National Engineering and Environmental Laboratory
P.O. Box 1625 MS 3875
Idaho Falls, ID 83415-3875
Phone: (208) 526-0283, Fax: (208) 526-1061
Hag@inel.gov

Greg Hulet

Mr. Hulet is the work package manager for the Unique Waste Work Package, which includes DOE's mercury mixed wastes. As such, he coordinates research, development, and technology deployment activities to ensure that all the wastes in the Unique category have a path for treatment and disposal. He has a Masters Degree in Chemical Engineering and ten years experience in waste management and pollution prevention. He also has considerable experience with Naval Nuclear Propulsion Plants. He has been a scoutmaster for 15 years, which, after watching scouts cook for that long, has made him an expert in unique hazardous wastes.

DOE Mercury Waste Treatment Demonstrations

Mercury has been used in Department of Energy (DOE) operations in a variety of applications. It has been used as a catalyst in nuclear fuel reprocessing, as shielding, and as a component of isotope separation processes. It is still being used in a number of facilities. Because of its widespread use, mercury contamination can be found at most DOE facilities. Efforts to clean up, treat and dispose the associated wastes are underway. However, for some DOE mercury wastes, until recently, no treatment processes were

available that had been demonstrated to be safe and effective in a radioactive environment. The DOE Mixed Waste Focus Area (MWFA) has been supporting research, development, demonstrations, and technology deployments to ensure that all mercury-contaminated waste can be safely treated and disposed. These activities have been divided into three main areas: amalgamation, stabilization, and separation. Subcategories of separation include removal of mercury from water, extraction from solid matrices, and gaseous emission control.

DOE supported the demonstration of two commercial mercury amalgamation processes. Both successfully amalgamated radioactive waste elemental mercury from DOE sites. The final waste forms met the Land Disposal Restriction for mercury, 0.2 ppm by Toxic Leach Characteristic Procedure (TCLP). Vapor pressure data for the waste forms are available.

The MWFA coordinated several commercial demonstrations for stabilization of mercury mixed waste with <260 ppm mercury. Allied Technology Group (ATG), Nuclear Fuel Services (NFS), and International Technologies (IT) performed bench-scale studies using surrogate waste with several species of mercury. ATG, NFS, and GTS Duratek demonstrated their respective processes on actual waste. In all cases the stabilized mercury met LDR limits. Reports covering each of these studies are available from the MWFA. Demonstrations are presently underway to treat >260 ppm mercury waste from Brookhaven National Laboratory (BNL). Sepradyne/Raduce is using their vacuum thermal desorption unit to extract mercury from the waste, while ATG, NFS, and BNL are using stabilization processes to treat the material. DOE is working closely with EPA on this project to acquire data that may support a change in the regulations for treatment of >260 ppm mercury-contaminated soils and sludges. BNL is the only group to have completed testing.

The DOE program for development of a process to extract mercury from solid matrices by non-thermal means is currently on hold because of funding cutbacks. The Polymer Filtration process dissolves mercury in shredded matrices and separates it from the solution using a complexing polymer. The process is ready for pilot-scale demonstration.

Oak Ridge conducted comparison tests of mercury sorbents on mercury-contaminated stream water from their East Fork Poplar Creek. ADA Technologies also tested their mercury sorbent process on the creek water with good results. Reports are available that summarize these two projects.

The MWFA is investigating continuous emission monitors for mercury but units are not available yet for commercial deployment. ADA has made progress in this area and in the area of sorption of mercury from gas streams.

Budget reductions have impacted work on the DOE mercury problems. Hopefully, funding will be available in fiscal year 2001 to bring the work to fruition.

Return and Recycling of Used High Intensity Bulbs for Recycling and Closed-loop Mercury Control

Lester Gress and Jeff Lord
Cleveland Fluid Systems Co.
PO Box 41070
Cleveland, OH 4414
Phone: (440)526-7070, Fax: (440)526-0770
lgress@aol.com

Return and Recycling of Used High Intensity Bulbs for Recycling and Closed-loop Mercury Control

Mercury is recognized as a highly toxic material and is stringently regulated in waste discharges. The majority of these discharges contain mercury in low concentrations limiting the control and recovery options. Wastes from a variety of industries generate wastewater containing residual mercury, including: lighting, medical, photographic, chloralkali, electronics and power generation.

The lighting industry has begun to address control and the reuse of mercury while they are trying to find substitute materials that adjust the electrical characteristics for the discharge lamp. One company has instituted a return of used high intensity lamps and the recovery of mercury from them. This program helps prevent mercury from entering into the eco-system. Some of the used and crushed glass is washed to insure the complete removal of mercury.

Typical treatment of wastewater requires multi-step processing ending in polishing steps that scavenge or trap residual mercury. These processing schemes result in added treatment costs and generate hazardous waste. A closed-loop mercury control/recovery system can reduce these treatment and disposal costs. The technology under development provides a means of accumulating sufficient mercury that recovery is possible and, at the same time, allows the minimization of the process wastewater by operating in a recirculating loop. Mercury is converted to its ionic form *in-situ* by chemical oxidation to improve solubility and is recovered electrolytically. The recovered mercury is relatively pure depending on the other contaminants present and potentially requires little additional processing before reuse.

Mercury Amalgamation Demos With the DOE

Clifton Brown
ADA Technologies, Inc.
8100 Shaffer Parkway, Suite 130
Littleton, CO 80127-4107
Phone: (303)792-5615, Fax: (303)792-5633
cliff.brown@adatech.com

Clifton Brown

Mr. Brown is currently the Vice President of Operations for ADA Technologies, Inc. -- a Denver-based technology R&D firm. Mr. Brown has 23 years of experience at Oak Ridge National Laboratory managing and performing R&D related to reactor fuel processing, coal conversion, and environmental processes.

Mr. Brown has B.S. and M.S. degrees in Chemical Engineering. Mr. Brown is also a Professional Engineer.

Recent Advances in Mercury Stabilization Technology

Since the early 1950s, mercury has been widely used throughout the DOE weapons complex. The legacy is contaminated solid waste, soils, and water. The main holders of mercury-contaminated waste are the Oak Ridge Reservation, the Idaho National Engineering and Environmental Laboratory, and the Savannah River Site.

Nationally, the largest categories of mercury-bearing wastes are sludges, soil, and debris. The Environmental Protection Agency subdivided mercury-contaminated solid wastes into three subcategories.

- Radioactively contaminated elemental mercury – treatment is amalgamation
- Low-mercury subcategory – treatment is stabilization
- High-mercury subcategory – treatment is thermal retort, followed by amalgamation if the recovered mercury is radioactively contaminated

ADA Technologies, Inc., has demonstrated and filed a patent for a process to handle radioactive elemental mercury. In recent studies this initial work has been extended to soil matrices that are contaminated with greater than 260 ppm mercury. Results derived from both of these studies will be presented and discussed.

Deployment of the Sulfur Polymerization and Stabilization Process as Applied to Mercury Contamination in Soils

Paul Kalb
Brookhaven National Laboratory
Environmental and Waste Management Group
34 Railroad St., Bldg. 830, P.O. Box 5000
Upton, NY 11973-5000
Phone: (631)344-7644
kalb@bnl.gov

Trevor Jackson
EnviroCare Utah, Inc.
46 West Broadway, Suite 116
Salt Lake City, UT 8410
Phone: (801) 532-1330, Fax: (801) 532-7512
tjackson@envirocareutah.com

Paul Kalb

Paul Kalb is a Senior Research Engineer at Brookhaven National Laboratory. He has a bachelor's degree in mechanical engineering from the State University of NY at Binghamton and a master's degree in nuclear engineering from Polytechnic Institute of NY. Paul has been employed at BNL for 20 years and has concentrated his efforts in the areas of hazardous/radioactive waste management, environmental restoration, and health and safety aspects of emerging energy technologies. Current responsibilities include Principal Investigator for programs on D&D and waste form development for DOE and industry. He has served as a member of several national technical support groups on Final Waste Forms for DOE and EPA, recently currently chaired a team that wrote a WASTECH volume on Stabilization/Solidification, is a member of the Program Advisory Committee for Waste Management Symposia, Inc., and has numerous patents and publications in the area of waste treatment and encapsulation.

Trevor Jackson

Dr. Jackson received his Ph.D. in Mechanical Engineering from Oklahoma State University in 1983. He spent two years as an Assistant Professor at the University of Maryland then progressed into industry. He was the site engineer at the solar energy plants located in the Mojave desert of Southern California, responsible for upgrades of existing plants. In 1988 he joined Science Applications International Corporation (SAIC) in San Diego providing assistance to the EPA in evaluating innovative technologies for the treatment of hazardous waste in the Superfund Innovative Technology Evaluation Program (SITE). He was Project Manager for the evaluation of many different technologies ranging from novel incinerators to bioremediation. In 1998 Dr. Jackson joined Envirocare of Utah, Inc., as Technology Development Manager. In this role he is responsible for reviewing and implementing treatment technologies for mixed low level waste at the Envirocare TSD facility in Utah. Dr. Jackson also upgrades performance of the existing stabilization, micro-, and macroencapsulation technologies.

Treatment of Elemental Mercury and Mercury Contaminated Soil and Debris by the Sulfur Polymer Stabilization/Solidification Process

Elemental mercury contaminated with radionuclides (mixed waste mercury) and mixed waste mercury-contaminated soil and debris, is a problem throughout the Department of Energy (DOE) complex. This presentation describes an innovative process developed at Brookhaven National Laboratory (BNL) and currently being commercialized at Envirocare of Utah, Inc., to immobilize mixed waste elemental mercury and mercury-contaminated soils and debris. The product is a monolithic solid waste form that is non-dispersible, will meet current and newly adopted EPA leaching criteria, and has low mercury vapor pressure. The BNL Sulfur Polymer Stabilization/Solidification (SPSS) process (patent pending) is a two-stage process that chemically reacts with mercury to form a product of low solubility and vapor pressure and then solidifies the product in a solid matrix to further reduce leachability and dispersion of contaminants. Waste forms containing as much as 33 wt% elemental mercury and as much as 60 wt% mercury-contaminated soil were formulated which successfully passed current Environmental Protection Agency Toxicity Characteristic Leaching Procedure (TCLP) criteria as well as the more stringent Universal Treatment Standard criteria that has been approved. In addition, the final waste form products exhibit extremely low leachability when subjected to long-term leaching, and significantly reduced vapor pressure compared with untreated mercury. Bench and pilot-scale development at BNL is complete and plans for commercial deployment at Envirocare's Clive UT mixed waste treatment facility are underway. The process may also be applied for direct and simple treatment of hazardous mercury streams as-generated or produced as secondary wastes from mercury separation technologies.

Commercializing a Safer Substitute for Mercury

James D. Rancourt, Ph.D.
NewMerc, Ltd
1872 Pratt Drive (MS 1260)
Blacksburg, VA 24060
Phone: (540) 951-2500, Fax: (540) 961-5778
info@newmerc.com
<http://www.newmerc.com>

James Rancourt, Ph.D.

Dr. James Rancourt obtained an undergraduate degree in Chemistry at the University of Lowell in Massachusetts. He earned a doctorate in chemistry, with an emphasis on analysis and preparation of electrically conductive plastics, from Virginia Tech. In 1987, Dr. Rancourt founded Polymer Solutions Incorporated, a company that provides innovative technical solutions to polymer and materials programs.

Dr. Rancourt led a research team to develop alternative materials for the mercury metal that is used in electrical switch applications, in 1992, at the request of the Virginia State Government. Dr. Rancourt's team now has four international patents and commercial products. He is the President of NewMerc, Ltd., a company devoted to producing reliable alternative materials to mercury metal for industrial and government applications.

Commercializing a Safer Substitute for Mercury

Mercury metal is a fundamental chemical element that has unusual properties: volatile, electrically conductive, reflective and liquid to low temperatures. Unfortunately, mercury metal, when handled or disposed of improperly, poses environmental and health risks. It is becoming increasingly important that mercury be replaced in industrial applications with a less toxic and reliable material. NewMerc, Ltd., has an exclusive, all-fields worldwide license to technology that offers a safe replacement for mercury in many applications.

This presentation will provide a brief description of the impetus for the nonmercury alloy development project, the research approach that was taken and the rationale for the technical solution that has been developed. The presentation will provide information about the composition of the alloy, its method of preparation and application areas. In addition, the properties of the NewMerc alloy, its MSDS sheet and questions remaining for the full-scale implementation of the patented material will be provided. A brief overview of the company structure will also be provided.

The Business of Mercury Pollution Prevention: Identifying Source Reduction Opportunities and Engineering Trade-Offs

Kenneth R. Stone
National Risk Management Research Laboratory
U.S. Environmental Protection Agency
26 W. Martin Luther King Drive
Cincinnati, Ohio 45268
Phone: (513) 569-7474, Fax: (513) 569-7111
stone.kenneth@epa.gov

Kenneth R. Stone

Kenneth Stone is the Engineering Trade-Offs Team Leader for EPA's National Risk Management Research Laboratory, based in Cincinnati, Ohio. Ken has been with the EPA for 18 years and has worked primarily in pollution prevention research with an emphasis on federal facilities and operations. Ken founded and managed the Life Cycle Engineering and Design Program, a cooperative venture with DoD to apply Life Cycle Engineering and pollution prevention methodology to industrial

systems. Ken's team is conducted research to advance the state of the practice of LCE and has completed several LCE case histories on both public and private products and operations.

The Business of Mercury Pollution Prevention: Identifying Source Reduction Opportunities and Engineering Trade-Offs

The demand for mercury in the United States is still growing or declining only slightly in a number of industrial sectors. These include electric lighting, electronic equipment, wiring devices and switches, measurement and control instruments, dental equipment and supplies, laboratory uses, and medical uses. About 190 tons of mercury were used by these sectors in 1997. While EPA is pursuing a number of voluntary initiatives within these industries, information on consumption, use, release and environmental impact is poor. Therefore, an assessment is underway to collect the data needed to identify the potential for source reduction across industry sectors. This assessment will determine in which areas emissions are large and difficult to measure. This assessment will incorporate collaborative activities with industry, including providing systems analysis tools such as Life Cycle Engineering (LCE) and Engineering Trade-Offs (ETO) to help industry determine the economic, energy, and environmental costs and benefits of management options.

The National Risk Management Research Laboratory (NRMRL) has initiated a Pollution Prevention Prioritization Assessment (P2PA), based on evaluation of the potential for source reduction of mercury use in the consumer sector, to identify major needs and opportunities for reduced use and releases. The P2PA will guide the development of at least two evaluations of pollution prevention approaches for mercury using life cycle analysis, and determine the reduction in adverse environmental impacts. The P2PA will also guide the selection of sector activities for evaluation of engineering trade-offs and input/output modeling.

ORD will use research innovative and emerging technologies for reducing reliance on mercury and mercury-containing products in these industries. This investigation will focus on source reduction opportunities. A compendium of technologies and technical solutions will be developed in order to inform the next step of the plan, prioritization.

A PBT Technology Information Clearinghouse

Frederic H. K. Booth
Waste Policy Institute
12850 Middlebrook Rd., Suite 250
Germantown, MD 20874-5244
Phone: (301) 528-1909, Fax: (301) 528-1970
fred_booth@gt.wpi.org

Kay Van der Horst
Associate Director, Environmental Security Programs
Waste Policy Institute
12850 Middlebrook Road, Ste. 250
Germantown, MD 20874-5244
Phone: (301) 528 - 1923, Fax: (301) 528 - 1971
Kvanderh@clark.net or Kay_vdh@gt.wpi.org

Frederic H. K. Booth

Mr. Fred Booth is the senior economist at WPI and has more than 25 years experience in leading economic, energy and environmental analysis programs. His experience includes analyses and optimization of energy, economic, and environmental system interactions; development of global climate change decision support tools and programs; development of environmental information systems architectures; systems analysis of local, regional, and national energy policy/regulatory issues; alternative fuels and electric utility demand forecasting; and technology diffusion analyses of advanced energy technologies. He has experience in evaluating the economic implications of proposed amendments to both RCRA and CERCLA. Additionally, his experience includes environmental technology cost analysis model development, econometric analyses, comparative and parametric life cycle cost modeling, innovative environmental technology cost-benefit analyses, and evaluation/demand forecasting for emerging technologies, particularly in energy and environmental markets.

Kay Van der Horst

Mr. Van der Horst is the Associate Director for Environmental Security Programs for WPI, a Virginia Tech owned not-for-profit organization. He is a specialist on domestic and international environmental security concerns with a particular emphasis on stakeholder involvement and risk communication. Currently, he is co-leading for WPI the development of EPA's new "PBT Information, Communication and Decision Support Clearinghouse". His responsibilities also include the development and implementation of Stakeholder Communication, Risk Communication, Training and Community Outreach Programs. Other programmatic areas focus on the development on risk management response and systems engineering. Prior to WPI Mr. van der Horst has worked in various capacities on environmental security issues for the University of Alaska-Fairbanks, Texas A&M University and various international institutions such as the United Nations and the European Parliament.

The EPA PBT Information and Communications Clearinghouse

Many EPA Offices individually address Persistent Bioaccumulative Toxics (PBTs) in varying contexts. The basic goal of the PBT Initiative is to identify and reduce risks to human health and the environment from current and future exposures to priority PBT pollutants and address them in an integrated manner. Implicit

in achieving EPA's objectives in the PBT initiative is effective, efficient, and focused information management in the context of PBT technical data, scientific data, and communications/outreach efforts. This presentation addresses the key aspects of developing and implementing an EPA/OPPT PBT Information and Communications Clearinghouse. This concept evolved from discussions with representatives of the various EPA Offices and programs represented on the Mercury Task Force.

The structural approach contemplated in the Clearinghouse is straightforward: Develop generic information management structures and strategies that are sufficiently flexible such that they can be adapted to accommodate potentially unique informational dimensions of any PBT, yet are consistent, comparable, and robust. Key features of this approach include: creating processes that support information flows into and from the Clearinghouse, and providing on-going opportunities for stakeholder information inputs in a dynamic information management environment. This approach directly contributes to ensuring cost effectiveness via economies of scale in managing multiple PBT data sets, and enhances the ability of the Clearinghouse to transparently provide user interfaces to similar information management activities at other federal agencies, universities, and research organizations.

The activities conducted in assessing the aspects of developing and implementing an EPA/OPPT PBT Information and Communications Clearinghouse will include:

- Defining the specific mission, objectives, and goal(s) of the PBT Clearinghouse;
- Defining/characterizing alternative PBT Information Clearinghouse structural approaches and the relative strengths and limitations of each structural alternative;
- Identifying preliminary opportunities for programmatic leverage;
- Identifying/characterizing the benefits of the PBT Clearinghouse;
- Identifying/characterizing existing information management activities that could either contribute to, or be considered competitive with, the OPPT PBT Clearinghouse;
- Identifying/characterizing the specific PBT Clearinghouse pre-implementation activities that will contribute to a successful, cost-effective, highly functional PBT Clearinghouse;
- Defining/characterizing stakeholder audiences (and their needs) for the mercury module of the PBT Clearinghouse;
- Identifying critical PBT Clearinghouse Quality Assurance issues, including information consistency, comparability, data validation and verification, and systems configuration;
- Identify international mercury information activities, including DOE/FETC, UAF, UNEP, AMAP, and the European Union; and

- Consideration of risks (technical, information management, performance, schedule) inherent in developing and implementing an activity such as the PBT Clearinghouse.

The EPA PBT Industry Technology Market Forum

Implicit in achieving the objectives of the EPA PBT initiative's guiding principles is the effective, efficient, and focused management of PBT information, scientific data, and communications/outreach efforts. These guiding principles include:

- Addressing problems on multimedia bases through integrated use of all EPA tools;
- Coordinating with and building on relevant international efforts;
- Coordinating with and building on relevant federal programs and agencies;
- Stressing cost-effectiveness (amount of PBT removed per dollar spent);
- Involving stakeholders;
- Emphasizing innovative technologies and pollution prevention;
- Protecting vulnerable sub-populations;
- Basing decisions on sound science; and
- Using measurable objectives and assess performance.

The subject of this presentation is assessing the key aspects of developing and implementing an EPA/OPPT PBT dynamic, stakeholder driven EPA/Industry Technology Market forum that is an integral element of an OPPT/PBT Information and Communications Clearinghouse which will directly contribute to implementing the guiding principles of the PBT Initiative. The EPA/Industry Technology Market Forum:

- Provides Regulatory Compliance Incentives for Industry by Providing Cost Savings Opportunities;
- Eases Regulatory Compliance Support by Providing Industry With Higher Production Efficiency Opportunities;
- Creates a Marketplace for Intercomparable/Verifiable Innovative Technologies;
- Fosters Development of Innovative Technology Developments by Expanding Hidden Technology Visibility; and
- Fosters Global Environmental Technology Improvement and Exchange.

The initial focus of the EPA Environmental Technology Market Forum will be mercury-related information, communications products and services. Though initially driven by a mercury focus, the EPA Environmental Technology Market Forum will be designed to accommodate a larger environmental technology market that addresses technology needs of all other PBTs.

WPI currently also supports the Department of Energy-National Energy Technology Laboratory (DOE-NETL) in the design, development and implementation of its Decision Support Center. The first information module in the Center focuses on DOE's coal combustion-based mercury data collection and analysis program. The NETL effort is also significantly driven by providing comparable technology solutions and information. This project represents both a unique opportunity for OPPT and the Mercury Task Force to apply real-time lessons learned from the NETL program, and additionally, leverage EPA and DOE mercury

program funds to improve the overall programmatic return on investment. Most significantly, the successful creation and implementation of the Environmental Technology Market Forum, in conjunction with the PBT Clearinghouse concept, represents the first of an on-going series of opportunities to leverage limited EPA resources.

Mercury Stabilization in Chemically Bonded Phosphate Ceramics

Arun S. Wagh, Ph.D.
Ceramist, Energy Technology Division
Argonne National Laboratory
9700 S. Cass Avenue
Argonne, IL 60439
Phone: (630)252 4295/5741, Fax: (630)252 3604
wagh@anl.gov

Arun S. Wagh

Dr. Wagh is working as a materials research engineer at Argonne National Laboratory and has a Ph.D. in physics. His expertise includes radioactive waste management, mineral waste management, and structural ceramics.

With his colleagues who are co-authors of this presentation, he developed chemically bonded phosphate ceramic program for radioactive and hazardous waste stabilization at Argonne National Laboratory, pioneered research on bauxite tailings (high volume residue from alumina refineries), directed projects related to utilization of greenhouse CO₂, and hot gas ceramic cross-flow filters at Argonne National Laboratory, and worked as consultant to alumina industries, that include, ALCOA, ALCAN, and Virgin Island Alumina Co.

Dr. Wagh was a recipient of the R&D-100 Award given by R&D Magazine in 1996 for 'Ceramicrete Binder', and the Pace Setter award by Argonne National Laboratory in 1997.

Mercury Stabilization in Chemically Bonded Phosphate Ceramics *

Mercury stabilization and solidification is one of the challenges for the conventional stabilization technologies. This is because of the stringent limits on leaching of its stabilized products that need to be enforced. In a conventional cement stabilization process, Hg is converted to its hydroxide at high pH which is not a very insoluble compound and hence sulfidation of Hg is considered to be a preferred route which converts it into an insoluble cinnabar (HgS). Unfortunately, efficient formation of this compound is pH dependent. At a high pH, one obtains more soluble sulfate of Hg, in a low pH range insufficient immobilization results due to

escape of hydrogen sulfide, while efficient formation of HgS occurs only in a moderately acidic region. This is the region (pH = 4-8) in which stabilization using Chemically Bonded Phosphate Ceramics is carried out.

This presentation will discuss this kinetics followed by our experience on bench stabilization of various U.S. Department of Energy (DOE) waste streams containing Hg in the Chemically Bonded Phosphate Ceramic (CBPC) process. This process was developed to treat DOE's mixed waste streams. It is a room-temperature-setting process based on an acid-base reaction between magnesium oxide and monopotassium phosphate solution that forms a dense ceramic within hours. For Hg stabilization, addition of a small amount (<1 wt.%) of Na₂S or K₂S is sufficient in the binder composition.

Here we discuss the Toxicity Characteristic Leaching Procedure (TCLP) results on CBPC waste forms of secondary waste streams generated from Hg-containing wastes such as combustion residues and Delphi "DETOXSM" residues. The results show that though the current limit on leaching of Hg is 0.2 mg/l, the results on the CBPC waste forms are at least an order lower than this stringent limit. This low leaching level provides robustness to the process and allows sufficient margin for the variability of Hg content in the waste. The efficient stabilization is attributed to chemical immobilization of Hg as cinnabar followed by its physical encapsulation in a dense matrix of the ceramic.

Using this process, Argonne-West has eliminated Hg-contaminated light bulbs from its inventory. These bulbs were slightly contaminated radioactively and hence this was a typical mixed waste stream. This presentation will provide a brief review on this work as an example of disposal of Hg-contaminated actual waste.

* Work supported by U.S. Department of Energy, Office of Technology Development, as a part of the Mixed Waste Focus Area, under Contract W-31-109-Eng-38, and Delphi Research, Inc., of Albuquerque, NM.

Characterization and Leachability of Stabilized Mercury-Containing Wastes

Linda Rieser
Academic Director
Accelerated Life Testing and Environmental Research (ALTER) Facility
University of Cincinnati
Cincinnati, OH
Phone: (513) 556-2060, Fax: (513) 556-3148
lrieser@uceng.uc.edu

Linda Rieser

Linda Rieser joined the University of Cincinnati in 1981. She served as Senior Research Associate from 1981 to 1991 and as Academic Director of UC's Accelerated Life Testing and Environmental Research (ALTER) Facility for the last 9 years. Her expertise includes the application of experimental methods to problems involving the solubility and mobility of hazardous and radioactive elements, the origin and remediation waters and soil, and the treatment of hazardous and radioactive wastes.

Characterization and Leachability of Stabilized Mercury-Containing Wastes

EPA's National Risk Management Research Laboratory (NRMRL) in collaboration with the University of Cincinnati established a research program supporting Agency actions on mercury; in particular, the potential revisions to the Land Disposal Restrictions for mercury-bearing wastes. Over the past year, research has been conducted on the characterization and leachability of several mercury waste forms. Wastes studied include mercuric sulfide sludges from several chemical plants, mercuric chloride catalyst used in the manufacture of vinyl chloride, surrogate mercuric chloride and elemental mercury wastes. This presentation describes characterization of the stabilized waste samples and analysis of leaching stability. The testing includes TCLP analysis and constant pH leaching tests to determine the potential mobility and stability of the mercury under simulated landfill conditions.

The work to be presented was performed by Paul Randall (EPA) and Paul Bishop, Haishan Piao, Renee Rauche, Linda Rieser, Makram Suidan, and Jian Zhang (UC).

Treatment of Wastes Contaminated with Mercury

Paul R. Lear, Ph.D.
IT Corporation
304 Directors Drive
Knoxville, TN 37923
Phone: (865) 694.7316, FAX: (865) 694.9573
plear@theitgroup.com

Paul R. Lear, Ph.D.

Dr. Lear has over 12 years experience in the treatment of hazardous waste with dewatering, soil washing and stabilization treatment technologies. He has experience in selecting and evaluating treatment alternatives and providing data for preliminary design activities and project equipment specifications. He has conducted research in the area of innovative stabilization systems, including systems for the stabilization of organic contaminants in hazardous wastestreams. Dr. Lear has also conducted research into the stabilization of metals, concentrating on arsenic, mercury, thallium, vanadium, antimony, and beryllium. He has extensive experience in the stabilization of hard-to-treat wastestreams, such as hazardous waste incinerator residues. Dr. Lear has hands-on experience with full-scale remediation activities and specializes in process troubleshooting. He has provided technical operational support to bioremediation, dewatering soil washing, stabilization, thermal, and wastewater treatment activities at remedial sites. He has also managed several pilot- and field-scale technology demonstrations.

Treatment of Wastes Contaminated with Mercury

This presentation will focus on the treatment of wastes contaminated with mercury. Four technologies (heavy metals bioremediation, surface decontamination, stabilization, and thermal desorption) applicable for the treatment of mercury wastes will be discussed, along with data from selected case studies.

Heavy metals bioremediation involves the stimulation of naturally occurring or augmented sulfur-reducing bacteria. These bacteria produce sulfuric acid and reduce the pH of the waste to below 2. Leaching of water through the waste removes the solubilized metals. The metals are then precipitated from the leach solution and sent for metals recovery or disposal.

Surface decontamination combines physical and chemical removal of contamination on the surface of debris such as concrete, block, and scrap metal. Extraction solutions containing chelants or acids are applied to the surfaces, allowed to react, and collected. Vacuum techniques are often applied to remove the extraction solution from semi-porous surfaces such as concrete. Multiple extractions are often required, especially on semi-porous surfaces.

Stabilization of mercury involves re-speciation of the mercury contamination to mercury sulfides. The chemistry required for re-speciation depends on the form of mercury in the waste. The solubility of mercury

sulfides is on the order of 10 mg/L. The mercury sulfides are then encapsulated in a cement matrix.

Thermal desorption involves the direct or indirect heating of the waste to volatilize the mercury. The temperature required for the volatilization depends on the form of mercury in the waste. The volatilized mercury is then condensed in the air pollution control system for recovery or disposal.

Case histories involve the application of thermal desorption and stabilization treatment technologies to mercury-contaminated wastes.

Treatment of Mercury-Bearing Wastes with Thermal Desorption Technology

David B. Malkmus
Applied Technologies Manager
SeptraDyne Corporation
7201 I-35 North
Denton, TX 76207
Phone: (940)243-8203, Fax: (940) 243-9089
Dmalkmus.sepradyne@iolt.com

David B. Malkmus

Mr. Malkmus received his degree (BS 1979) in Chemical Engineering from Clemson University with specialization toward Environmental Engineering. He has over 20 years experience in the design, startup operation and project management of waste processing systems used in the commercial and energy industries including the Department of Energy and commercial nuclear power plants. Mr. Malkmus has designed large scale, proprietary waste treatment systems incorporating advanced water processing and state of the art waste minimization technologies. He has published several technical papers regarding technology advances through EPRI, US DOE and the International Water Conference.

High Vacuum Rotary Retort for the Recovery of Products and the Minimization of Wastestreams

At a Westinghouse subsidiary, Scientific Ecology Group, Mr. Malkmus served as a fellow engineer responsible for the evaluation, development, and deployment of new technologies for waste treatment applications in addition to serving as a project manager in the Operations Department. Prior to that, he held engineering and operation management positions with VECTRA Technologies, the SCANA Corporation: VC Summer Nuclear Power Plant and the NUS Corporation.

The SeptraDyne Corporation has commercialized an extremely cost-effective process for removing and recovering constituents having boiling points below 800°C. The process further provides a highly efficient reduction in the volume of any remaining non-volatilized media. The process material is indirectly heated within a rotating vessel under a high vacuum inert environment. The constituents of concern are volatilized and diffused from the feed material through the off-gas treatment train. Volatile constituents are condensed to liquid through an advanced impinger, chill water system. By operating under high vacuum, the material boiling points are significantly reduced thus enabling the ease of product recovery at lower operating temperatures. There is little decomposition of products due to thermal energy. Since the desorption and product recovery process is performed in an oxygen-free inert environment, there is no generation of furans and dioxins as well as any products of incomplete combustion. All retort off-gases are condensed to liquid eliminating the potential release of toxic substances to the atmosphere and thus permitting the recovery of the constituents for beneficial use. In addition, secondary waste streams are not produced because a steam or gas stripping media is not required to remove and transport chemicals from the processed material.

This paper will provide an overview of SeptraDyne vacuum desorption system(s) and outline the technological advances of the indirectly heated high vacuum retort. Also included are the results of several commercial and DOE applications for the separation of mercury from previously classified waste stream sources the minimization of waste sources and the near complete destruction of furans and dioxins.

Permanent Mercury Disposal in Sweden

Kristina von Rein
Principal Technical Officer
Section for Chemicals Control
Swedish Environmental Protection Agency
S-106 48 Stockholm, Sweden
Phone: +46(0)8-6981127, Fax: +46(0)8-6981222
Kristina.von-rein@environ.se

Kristina von Rein

Ms. von Rein has been with the Swedish Environmental Protection Agency since 1990, and is the project leader for the Governmental Assignment that includes both an Action Programme for more efficient collection of used goods and products containing mercury and preparation of a proposal for final disposal in Sweden of mercury-containing waste.

Ms. von Rein has a M.S. degree in chemical engineering.

Permanent Mercury Disposal in Sweden

Phase-out of mercury

Mercury is currently being phased out by means of various bans on the use of goods and products containing this metal, the target year being 2000. Also, the export of mercury as a residual product has been prohibited since July 1, 1997. Exports of mercury waste for reprocessing and reuse abroad is not a feasible alternative, at the same time as use of mercury in Sweden is being phased out. The Agency believes that capacity for disposal of mercury-containing waste should exist within the country.

Mercury is one of the most toxic of all pollutants. The burden of mercury on our environment must be reduced since every addition is undesirable. The Swedish EPA believes that it is our generation that must reverse the trend in order to create a healthy living environment for future generations. The question of how to store waste containing mercury ultimately concerns finding a way of detoxifying our society.

Disposal of mercury-containing waste

Large quantities of discarded goods are currently in storage pending a solution. Large amounts of waste are also stored in industry, either temporarily or at sites which do not meet long-term environmental safety requirements. The Swedish EPA considers this situation to be untenable. It is therefore essential to find a method for the terminal storage of mercury.

In December 1997, the Swedish EPA presented a report, concluding several years of investigations, to the Swedish government with the conclusion that disposal of waste containing mercury demands a tailor-made solution. The Swedish Environmental Protection Agency believes that mercury-containing waste should be disposed of in such a manner that the mercury leaks to the external environment as little as possible, viewed in a long-time perspective.

Deep storage rock - the best alternative

Alternative solutions have been compared, with a view to finding the form of terminal storage which best fulfils stringent environmental requirements. The alternatives compared are high-quality surface storage, shallow storage in rock and deep storage in rock. These options differ in philosophy and the way in which the surrounding environment must be protected against emissions.

The EPA considers that deep storage in rock is the safest method of storage in the long term, since it is the solution most in harmony with the environment; i.e., nature is used as a barrier and a buffer. The surrounding bedrock will protect the functionality of the storage facility for thousands of years or even longer. This solution can and should also be accompanied by technical measures to further reduce the risk of future emissions and to compensate for our lack of knowledge about the long-term processes governing the dispersal of mercury.

Sub-Seabed Emplacement: Long-Term Ultimate Disposal of Mercury Wastes in Geologic Formations on Land

D. R. (Rip) Anderson, Ph.D.
Sandia National Laboratories
P.O. Box 5800, MS-0779
Albuquerque, NM 87185
Phone: (505) 284-4600
drander@sandia.gov

D.R. (Rip) Anderson, Ph.D.

Dr. Anderson has 39 years of experience at Sandia National Laboratories and currently is the Project Manager for Sandia activities supporting the Waste Isolation Pilot Plant. Dr. Anderson's responsibilities include: technical analysis, code development, quality assurance, testing, field and laboratory data analysis, geotechnical and geochemical analysis, and incorporating the above into performance assessment calculations for the Waste Isolation Pilot Plant.

Dr. Anderson is an internationally recognized expert in risk and performance assessment. As manager of the WIPP Performance Assessment Department, Dr. Anderson led the construction and preparation of performance assessment analysis for a compliance certification application to EPA which has led to the opening of the first deep geological repository for radioactive wastes in the U.S. Dr. Anderson also has led numerous waste disposal and management efforts, including, but not limited to, the Sub-Seabed High Level Waste Project, the FUSRAP Disposal Program, and the Decommissioned Nuclear Submarine Program.

Dr. Anderson has authored and co-authored more than 50 publications and reports dealing with waste disposal, performance assessment and risk assessment. Dr. Anderson is also the editor of the *Radioactive Waste Management Journal*.

Dr. Anderson earned a B.S. in Chemistry (1957) from Idaho State University, and a Ph.D. in Theoretical Organic Chemistry and Chemical Oceanography (1961) from Oregon State University.

Land-Based Geologic Emplacement of Mercury Wastes

In 1979, Congress authorized the U.S. Department of Energy to build a research and development facility - the Waste Isolation Pilot Plant (WIPP) - to demonstrate the safe disposal of defense nuclear wastes containing transuranic radionuclides. The WIPP, located near Carlsbad, NM, was opened as the world's first nuclear waste repository and received its first shipments of transuranic wastes in March 1999.

The overall process of assessing whether or not a waste disposal system meets a set of performance criteria is known as a Performance Assessment (PA). The WIPP PA, conducted by Sandia National Laboratories,

provided important input to decisions on the safety of a plan of action using a detailed procedure and scientific knowledge. For radioactive wastes, a computationally demanding set of risk-based performance criteria was specified by the U.S. Environmental Protection Agency (EPA). These were quantitative criteria that specified probabilistic limits that had to be met for the first 10,000 years of operation of a nuclear waste facility. The WIPP PA group developed a suite of models to predict future behavior of the facility. The physical, chemical, and geological processes that determined the behavior and evolution of the WIPP site were complex and highly nonlinear. The PA models that describe the processes are complex and technically sophisticated, and can be used to study the feasibility of the disposal of non-radioactive environmental contaminants with infinite half-lives, such as mercury product wastes, in a land-based repository.

Mercury-Sniffing Dogs: The Swedish Experience

Kjell Avergren
Environmental Dogs' Manager
The Dog Training Centre in Sollefteå, Sweden
+46 302 326 79
kjell.avergren@swipnet.se
<http://www.humanitydog.com>

Kjell Avergren

Mr. Kjell Avergren has worked with environmental issues on both a governmental and a consultant level (local, regional and national) since 1980. He has lead the four Mercury Tracing Dogs projects within the Swedish EPA's mercury collecting program.

The Dog Training Centre in Sollefteå and Mercury Decontamination.

Mercury - An environmental problem

Mercury is one of the world's most serious pollutants. One way to protect the environment is to remove the mercury and deposit it in safe storage. The Swedish Parliament has concluded that mercury plays no part in the natural world and the use of mercury should be phased out by the year 2000. In several mercury-collecting projects with the Swedish EPA, the Dog Training Centre in Sollefteå showed that it is possible to obtain low-cost, successful, and rapid results using mercury tracing dogs. The strategy was to work together with many different actors, rather than using new or more regulations. The outcome was remarkable..

Mercury Dogs - The Cost-Effective Solution

In laboratories and chemical store cupboards, in hospitals, doctors' surgeries and dentists' consulting rooms and throughout industry, mercury can be found in sinks, drains and sewage systems. The Dog Training Centre now offers a mercury tracing service using sniffer dogs (The German Shepherd mr Froy and the labrador mr Ville Sigmund) and dog handlers. The service enables the cost-effective recovery of the mercury and prevents it from being dispersed in the environment through refuse or in the sewage system. Using sniffer dogs benefits the environment and the customer's bank balance. Tests have shown that using the dogs protects the environment, saves time and money, and generates goodwill.

More than 3,000 kg of mercury were collected from more than 1,200 schools, 20 universities and many hospitals taking part in different Swedish EPA projects. A number of doctors' surgeries, dentists' rooms, laboratories and business premises also participated in the projects. The dogs traced hidden mercury in sinks and floors in many thousands of buildings. German Shepherd mr Froy and labrador mr Ville Sigmund from the Dog Training Centre in Solleftea, Ltd., saved at the same time up to 3-3.5 million U.S. dollars in reduced decontamination costs. On average 5,300-8,800 U.S. dollars in clean-up costs were saved each working day, resulting in a short pay-off time.

The dogs' achievement has attracted positive publicity from television, radio, newspapers, magazines and on the Internet in both Sweden and abroad. They have been the subject of more than 2,000 items, including 90 television programmes.

The Dog Training Centre is part of the Iris Group, owned by the Foundation of the Visually Impaired, and the company's profit benefits the visually impaired. Humanity Dog trains guide dogs for the blind and breeds dogs to detect drugs, mould, PCB, oil, fire and mines. As part of its constant effort to improve the environment, the Centre has joined the "Green Trade network", established by the Swedish Trade Council.

Mercury Source Reduction and Recycling in Electrical Products

Eric (Ric) Erdheim

Senior Manager/Government Affairs, National Electrical Manufacturers Association

Executive Director, Thermostat Recycling Corporation

National Electrical Manufacturers Association (NEMA)

1300 North 17th Street, Suite 1847

Rosslyn, Virginia 22209

Phone: (703) 841-3249, Fax: (703) 841-3349

ric_erdheim@nema.org

Eric (Ric) Erdheim

Ric Erdheim is Senior Manager for Government Affairs at the National Electrical Manufacturers Association. He represents electrical manufacturers on environmental, occupational health, consumer product safety, and

fire safety issues. He also serves as the Executive Director of the Thermostat Recycling Corporation, an organization formed by the major thermostat manufacturers to operate a wholesaler take-back program for mercury switch thermostats.

Mr. Erdheim spent ten years as a Congressional aide, most of that time as Environmental Legislative Assistant to Senator Frank R. Lautenberg of New Jersey. Mr. Erdheim played a significant role in enactment of the ozone transport and air toxics provisions of the Clean Air Act Amendments of 1990, the Pollution Prevention Act, the Ocean Dumping Ban Act, and the Mercury Containing and Rechargeable Battery Management Act.

Mr. Erdheim graduated from the University of Pennsylvania with a BA in Economics and the George Washington University Law School.

Mercury Source Reduction and Recycling in Electrical Products

Manufacturers have used mercury in batteries, lamps and thermostats. Each industry has adopted different approaches to reducing environmental exposure to mercury that reflect the unique characteristics of the product.

In the 1980s, battery manufacturers used over 1,000 tons of mercury a year, mostly to make alkaline batteries. In response to environmental concerns, the industry developed alternatives to mercury in virtually all batteries. As a result, the only consumer batteries manufactured today that contain any mercury are button cell batteries. With the phase-out of mercury use by 1993, mercury from alkaline batteries in the waste stream has dropped from 10,000 PPM to less than 300 PPM. This level will decline by 50% every two years. This significant decline has been partially responsible for the declines in mercury levels from incinerators.

Lamp manufacturers have reduced the average mercury level in four foot lamps from 48 mg in 1985 to 11.6 mg in 1999. As a result, mercury contained in lamps has dropped significantly. More importantly, use of mercury-containing lamps results in a net decrease in mercury because of the energy efficient nature of the lamps as contrasted with no mercury but energy inefficient incandescent bulbs.

Manufacturers cannot reduce the amount of mercury used in mercury switch thermostats. The average mercury level in these thermostats is 3-4 grams. To address the problem of disposing of a product with such relatively high levels of mercury as compared to lamps, manufacturers have established the Thermostat Recycling Corporation to recapture mercury-switch thermostats. In the first eighteen months of operations in nine states, the TRC has recovered 270 pounds of mercury. This program works because of the unique characteristics of thermostats and is not necessarily a model for other products.

These examples indicate that manufacturers use mercury in a wide range of products for different purposes. The products differ in: units sold, mercury levels, size and product composition, users (businesses/specialized

installers/homeowners) and other factors. Because of these differences, issues involved in waste management vary for each product necessitating different approaches.

DSCP Buying Green

Anthony Armentani
Program Manager, Lighting Products
Defense Supply Center Philadelphia
700 Robbins Ave
Philadelphia Pa. 19111-5096
Phone: (215) 737-8047, Fax: (215) 697-9093
aarmentani@dscp.dla.mil

Anthony Armentani

Anthony Armentani is currently the Program Manager for Lighting Products at the Defense Supply Center Philadelphia. In this position Mr. Armentani is responsible for leading a team of associates in the acquisition, inventory management, technical and quality support initiatives in the management of over 60,000 commercial, non-commercial and military unique lighting items.

Mr. Armentani has over twenty years of federal service, all with the Defense Logistic Agency. He started his career at the Defense Industrial Supply Center (DISC) in Philadelphia as an Equipment Specialist and quickly moved up to Team leader, Supervisor and Branch Chief of the Miscellaneous Hardware and Physical Security Equipment unit in the Technical Operations Directorate. Mr. Armentani spent four years, as the Technical Data Manager at DISC where he was responsible for the acquisition, management and distribution of all the technical data required for competitive procurements at the center. Mr. Armentani spent two years on the Commanders staff at DISC, reengineering the work processes and participated in the development of DSCP's Innovative Logistic Support units that have allowed for a strong customer focus.

Mr. Armentani is a graduate of Rowan University in Glassboro, NJ.

Buying Green

The DSCP presentation will cover the methods and guidelines utilized by the DSCP Lighting Team in the acquisition and support of energy-efficient low-mercury lighting products. The presenter will discuss the DSCP/DLA customer commitment, The advantages and related savings in the use of low- mercury energy-efficient lamps and the projects and partnerships that we nurture and develop to ensure widespread energy-efficient lighting use throughout the federal sector. The briefing will identify Energy and environmental guidelines used in the acquisition of energy-efficient products, various types of low-mercury lamps available through DSCP and what new technologies are on the horizon for federal energy users. The presenters will

also discuss the different ways to research, select and order these energy-efficient low-mercury products from DSCP. Primary presenter for DSCP will be Tony Armentani Program Manager–Lighting.

EPA/AHA Agreement: Reduction of Mercury Wastes from Hospitals/Health Care Facilities

Chen Wen
Program Analyst
Pollution Prevention Division, Office of Pollution Prevention and Toxics
U.S. Environmental Protection Agency
401 M Street, SW (MC-7409)
Washington, DC 20460
Phone: (202) 260-4109, Fax: (202) 260-0178
wen.chen@epa.gov

Chen Wen

Chen is currently serving as a team member of the EPA Hospitals for a Healthy Environment (H2E) project, and staffs a number of different workgroups associated with H2E. Prior to working on H2E, Chen served in a number of diverse posts throughout the EPA, including:

- Program manager of the Environmental Justice Through Pollution Prevention Grant Program;
- Program manager of the Pollution Prevention and Insurance Project;
- Vice President Gore's Task Force for Government Reinvention;
- Agency Task Force for Contracts Management Reform.

Prior to joining the EPA, Chen obtained his Bachelor of Arts degree in Political Science from the University of Washington, and his Master of Arts degree in Public Policy Studies from the University of Chicago.

EPA/AHA Agreement: Reduction of Mercury Wastes from Hospitals/Health Care Facilities

EPA's Voluntary Agreement with the American Hospital Association and Its Implications on the Need for Agency Standard for the Disposal of Mercury According to EPA's Mercury Report to Congress, the healthcare industry is the 4th largest source of mercury release. The mercury release eventually find its way into the food chain, and back to humans.

The voluntary agreement between the American Hospital Association - which represent some 85 percent of all healthcare facilities in the United States - and the EPA outlines a number of goals. One of

the stand-outs is to “virtually eliminate” mercury-containing waste by 2005.

Mercury Content of Products Commonly Used by Boston Area Hospitals

Kevin McManus
Toxic Reduction and Control Department (TRAC)
Massachusetts Water Resources Authority
5313 38 St., NW
Washington, DC 20001
Phone: (202)362-6034, Fax: (202)362-6632

Kevin McManus

Mr. Kevin McManus is the Director of the Toxic Reduction and Control Department (TRAC) of the Massachusetts Water Resources Authority.

Mr. McManus is responsible for implementation of the MWRA's Industrial Pretreatment Program. MWRA currently regulates approximately 1,100 industrial and commercial dischargers in order to control the loadings of heavy metals and organic pollutants to MWRA's new treatment plant on Deer Island. TRAC also works with trade organizations, municipalities and other agencies to reduce toxics from a wide array of non-industrial sources such as hospitals, laboratories, photoprocessors, dental facilities and automotive facilities.

Prior to coming to the MWRA in 1993, Mr. McManus worked for seven years with Metcalf and Eddy, Inc., managing the environmental compliance programs for numerous private and public construction projects. He also worked as General Manager for Offshore Devices, Inc., a marine engineering firm specializing in the manufacture and use of offshore oil spill cleanup equipment.

Mr. McManus has an undergraduate degree in Marine Policy from the University of Rhode Island, a Masters degree in Marine Resource Management from the University of Washington, and a Master of Business Administration degree from Boston University.

New Strategies for Reducing Mercury Discharges from Health Care Facilities

The five-year MWRA/Hospital Mercury Workgroup is a cooperative effort between the Massachusetts Water Resources Authority (MWRA) and Boston-area hospitals and medical facilities to reduce the discharge of mercury-containing products from hospitals to the sewer system. This workgroup identified mercury in many products that have commonly been used in hospitals and other medical facilities, such as blood test reagents and cleaning products. The workgroup has actively researched mercury-free alternatives to many of these products, and developed a mercury products

database which is available to area hospitals and other interested parties.

The workgroup also:

- Developed standards for replacing piping where mercury can accumulate over time;
- Prepared guidance documents for industries detailing mercury compliance problems;
- Assessed loadings of mercury from industrial dischargers in the MWRA sewer service area; and
- Conducted pilot-scale testing of promising mercury pretreatment systems.

A key factor in gaining the cooperation of facilities in the workgroup was MWRA's Mercury Safe Harbor Program. Under this program, MWRA will not escalate enforcement (beyond enforcement orders) against companies that have non-compliant mercury discharges, provided they actively participate in the program and demonstrate progress in reducing their mercury discharges. Under this program, the MWRA has divided its non-compliant mercury dischargers into two groups. Group 1 consists of sewer users whose discharge contains 0.004 mg/l or less of mercury; Group 2 consists of sewer users whose discharge contains more than 0.004 mg/l of mercury. Facilities that operate outside the safe harbor will be subject to escalating enforcement including monetary penalties.

To date, this cooperative effort has resulted in a significant decrease in mercury concentrations from the facilities permitted by the MWRA in the metropolitan Boston region. The 29 major hospitals and medical centers (representing 55 individual sampling locations) were a major source of mercury from MWRA's permitted users. Since 1995, 77% of these sampling locations have achieved compliance (1 part per billion or less), and only 9% remain above 4 parts per billion on a consistent basis. Average mercury discharge concentrations from these hospitals dropped from 22 ppb in 1994 to 2 ppb in 1999.

Eliminating Non-Essential Mercury Uses

Michael T. Bender
Mercury Policy Project
1420 North St.
Montpelier, VT 05602
Phone: (802) 223-9000, Fax: (802) 223-7914
MTBenderVT@aol.com
<http://www.mercurypolicy.org>

Michael T. Bender

Michael Bender is a consultant to the Mercury Policy Project, a small, nonprofit enterprise dedicated to reducing human exposures to mercury and the virtual elimination of anthropogenic mercury releases. The Project identifies strategic opportunities and works collaboratively with business, government and nongovernment officials toward attaining its goals.

Michael has over 10 years experience in municipal hazardous waste management and has focused on mercury for the past several. From 1995 to 1997, Michael worked to secure the release of the Mercury

Report to Congress and since then has provided input on the Universal Waste Rule and the Mercury-Containing Lamps Rule, the New England Governors/Eastern Canadian Premiers Mercury Action Plan, the North American Regional Action Plan on Mercury and the Agency for Toxic Substances and Disease Registry's mercury reference level.

Michael has a Bachelor of Arts in General Studies and a Masters of Science in Resource Management and Administration from Antioch New England.

Phasing Out Thermometers and Other Non-Essential Mercury-Containing Products

The top priority of federal, tribal and state waste hierarchies is source reduction, with special attention paid to eliminating substances in products, like mercury, when they are found to present some of the most profound risks to human health, wildlife and the environment. For mercury-containing products, then, whenever viable, environmentally sound and cost-effective alternatives are identified that contain no mercury, they should become the preference of government procurement programs and strongly supported as the preferred societal choice. For this to occur effectively, non-essential mercury-containing products must be “virtually eliminated” over time by phasing out their manufacture, import and sale. As an interim step, existing products should be collected and properly managed to prevent the haphazard release of mercury indoors or into the environment. There are currently a number of initiatives where both voluntary and mandatory phase-outs of mercury-containing products are being carefully considered, developed or implemented. This paper will present several case study examples.